Roll No.

Total Pages: 4

10308/NH

C-5/2110

ORGANIC CHEMISTRY-B

Semester-V

Syllabus-(Dec-18)

Time allowed: 3 Hours] [Maximum Marks: 26

Note: The candidates are required to attempt two questions each from Section A and B carrying 4 marks each and the entire Section C consisting of 5 short answer type questions carrying 2 marks each.

SECTION-A

- Explain the principle of NMR Spectroscopy. 1.
 - What do you understand by equivalent and non equivalent protons? Illustrate with suitable example. (4)
- 2. How will you distinguish the following pair of compounds on the basis of PMR spectroscopy?
 - (a) Propene and cyclopropene
 - (b) 1,1 dibromoethane and 1,2 dibromoethane

[P.T.O.

standard? (4) Give account oi following: (a) Hypsochromic shift

What is chemical shift? What are the scales of

measurement with TMS as reference

3.

- (b) Chromophore
- (c) Auxochrome
- (d) Lambert's law
- (ii) What is the effect of H-bonding on absorption frequencies of Alcohol and Amine? (4)
- On the basis of Woodward fischer rule calculate for following compounds:
 - Methyl vinylketone
 - 2-cyclopentane

(4)

SECTION-B

On the basis of IR Spectroscopy how acid, 5. chloride, anhydride, esters, arid amides can be distinguished from one another.

- (ii) How would IR Spectra of following pairs of compounds differ:
 - (a) Acetone and Ethanol
 - (b) Acetic Acid and Methanol (4)
- 6. Discuss 1, 2 and 1, 4 addition of Grignard's reagent to , unsaturated ketone. How does steric hindrance affect the course of these reactions? (4)
- 7. (i) Why does Sulphur prefer to form p -d bond.
 - (ii) Give two methods for synthesis of thioethers from alkylhalide. (4)
- 8. How are 1°, 2°, 3° alcohol prepared from alkyl lithium? (4)

SECTION-C

- 9. (i) What is shielding and deshielding of protons?
 - (ii) What factors affect chemical shift.
 - (iii) Explain briefly types of electronic transitions.
 - (iv) What is strecker reaction give its chemistry.

- (v) Write notes on:
 - (a) Reformats by reaction
 - (b) Simmon Smith reaction. $(2\times5=10)$