703/MH

C/2050 PHYSICAL CHEMISTRY (COMMON FOR B.Sc. BIO TECHNOLOGY)-III SEMESTER-VI

	TIME ALLOWED 2 Hrs	M.M.26
Special .	Explain Raman Effect on the basis of polarizability of molecules. How is the rota	tional
	spectrum observed experimentally?	4
II.	a) Explain the relative intensities of the lines obtained in a pure rotational spectrum.	
	b) Are the lines in the rotational Raman spectra equally spaced? Elaborate your answ	ver.
		2, 2
III.	a) What are the selection rules for transitions in electronic spectroscopy of molecules	s?
	b) Briefly explain Franck-Condon Principle.	2, 2
IV.	a) Derive Bragg's equation for X-ray Diffraction by crystals.	
	b) At What angles will X-rays of wavelength 1.542x10 ⁻¹⁰ m undergo first orde	r and
	second order reflections by planes separated by 3.54x10 ⁻¹⁰ m.	2, 2
V.	Describe the theory of photochemistry of vision and color.	4
VI.	State and explain first and second law of Photochemistry. What do you understand by	y
	'one Einstein' of energy? How is its value calculated in CGS units and in SI units?	4
II.	a) Calculate the value of Einstein in kiloJoules for orange light with $\lambda = 600$ nm.	
	b) Draw the Jablonski diagram depicting various processes.	2, 2

- VIII. What is Photosensitization? Illustrate and explain with three examples along with their mechanism.
 - IX. a) The crystallographic axes of a unit cell are \vec{a} , $\vec{b} \& \vec{c}$. Draw a plane of which Miller indices are (221). Can there be more than one plane with the same Miller indices.
 - b) Define Auger effect. What is the main advantage of powder method over Bragg's method? How is interplanar spacing calculated from X-ray diffraction pattern of a powder sample?
 - c) On the basis of mechanism, how can you justify tat quantum efficiency of photolysis of HI is 2.
 - d) Out of CO₂, C₂H₆, H₂O, NO, N₂O which will exhibit pure rotational spectra?
 - e) In the electronic band spectrum, which transition out of $n \to \pi^*$ and $\pi \to \pi^*$ will give greater intensity and why? What happens if an acid is present? 2x = 10